



ARTICLE

Gas Production and Reservoir Settlement in NGH Deposits under Horizontal-Well Depressurization

Lijia Li, Shu Liu, Xiaoliang Huang* and Zhilin Qi

School of Petroleum Engineering, Chongqing University of Science and Technology, Chongqing, China

*Corresponding Author: Xiaoliang Huang. Email: 2009022@cqu.edu.cn

Received: 15 September 2025; Accepted: 09 January 2026; Published: 06 February 2026

ABSTRACT: Identifying geohazards such as landslides and methane leakage is crucial during gas extraction from natural gas hydrate (NGH) reservoirs, and understanding reservoir settlement behavior is central to this assessment. Horizontal wells can enlarge the pressure relief zone within the formation, improving single-well productivity, and are therefore considered a promising approach for NGH development. This study examines the settlement response of hydrate-bearing sediments during depressurization using horizontal wells. A fully coupled thermal, hydraulic, mechanical, and chemical (THMC) model with representative reservoir properties (Shenhu region in the South China Sea) is presented accordingly. The simulations show that lower production pressures, while increasing gas output, also intensify formation settlement. The maximum difference in settlement between the lowest and highest production pressures reaches 0.016 m, contributing to more pronounced differential subsidence. Optimal well placement, specifically targeting a low-saturation hydrate zone containing free gas and situated adjacent to a high-saturation hydrate layer, markedly improves both gas production rate and cumulative yield, while reducing overall settlement and limiting changes in effective stress.

KEYWORDS: Hydrate bearing sediment; horizontal well; depressurization; reservoir settlement

1 Introduction

Nature gas hydrate (NGH) is an ice-like crystal that mainly exists in Oceanic continental margin and permafrost region [1]. Global reserves of NGH are estimated to be twice those of conventional fossil fuels, making them a promising alternative energy source for the future [2]. Various techniques have been proposed for gas production from NGH reservoirs, including depressurization [3], thermal stimulation [4], chemical inhibitor injection [5], and gas replacement [6]. Among these methods, chemical inhibitor injection has been extensively studied for natural gas pipeline transportation and well drilling, but has received relatively limited attention for NGH production due to its potential environmental risks [7]. The gas replacement method offers dual benefits of CO₂ storage and NGH production, along with the added advantage of reservoir repair [8]. CO₂ can also enhance production by forming a hydrate cap or through displacement as a mixed gas with N₂ [9]. However, the low replacement rate makes it difficult to meet the commercial production demands [10]. Thermal stimulation methods [11–13] are designed to supply the heat required to maintain equilibrium temperature during the rapid decomposition of NGH. Nevertheless, field tests have revealed significant energy losses associated with thermal stimulation methods due to the low permeability [14] and thermal conductivity [15] of the NGH reservoirs. In general, depressurization is widely regarded as the most economical method for NGH production, as it does not necessitate additional

energy input. Both numerical simulation at the field scale and field trials have shown their potential commercial application [16].

Previously, field trials primarily employed vertical wells to test gas production from the NGH reservoir. To obtain a high gas production rate from vertical wells, a significant pressure differential is applied to accelerate the NGH decomposition rate [17]. However, large pressure drawdown can trigger Joule-Thomson effect combined with the endothermic of NGH decomposition, causing a rapid decline in reservoir temperature and resulting in the reformation of NGH and ice [15]. In comparison, horizontal wells can expand the pressure relief zone in the reservoir, which facilitates gas production from individual wells even at a lower pressure differential. This approach has shown promise, as demonstrated by the second NGH production field trial in the South China Sea [18].

Beyond economic considerations, safety risks represent a critical constraint on NGH exploitation. Marine NGH reservoirs are typically characterized by shallow burial depth and weak diagenesis properties [19]. In porous sediments, NGH often serves to cement skeletal particles as the primary binding agent [20]. The dissociation of NGH weakens the mechanic properties of the reservoir, leading to reservoir settlement, which may cause wellbore damage, submarine landslide [21], and methane leakage incidents [22]. Therefore, it is necessary to understand the geomechanical response characteristics during NGH production to evaluate the stability of disturbed formations. The mechanical properties of NGH reservoirs have been widely studied, revealing that their strength and stiffness depend on the NGH saturation [23], distribution morphology [19] and the characteristic of the sediment skeleton particles [24]. Numerous studies show that hydrates have less influence on the strength of NGH bearing sediments when the NGH saturation is less than 10% [25], then its strength increases significantly with rising hydrate saturation [26]. Microscopic studies indicate that this phenomenon is attributed to the distribution morphology of the hydrate in porous media. Hydrate is more likely to fill the pores space when its saturation is below 10% [21], then gradually cements the sediment particles with continues increase in saturation [27]. The earth pressure coefficient of NGH is mainly controlled by the sediment skeleton particles at high stresses [28]. Hydrate exploitation leads to the deterioration of the mechanical properties of the reservoir, and the reservoir is bound to settle under the stress of the overlying formation. Several Studies show that the maximum strain occurs at the reservoir boundary directly below the well, with values reaching up to 0.6% [29]. These deformations and settlements cause uneven horizontal *in-situ* stress on hydrate-bearing reservoirs, which will cause wellbore wall instability during hydrate production [30].

Several software packages are available for the numerical simulation of hydrate reservoirs, such as TOUGH+Hydrate [31] and COMSOL [32]. TOUGH+Hydrate can be coupled with FLAC3D to simulate the geomechanical behaviors of hydrate reservoirs. However, mesh generation remains complex, the numerical simulation of hydrate reservoirs, most studies focused on the geomechanical response characteristics in the NGH production process by vertical wells with field scale [33–35]. Seldom considers the impact of heterogeneous hydrate distribution on reservoir settlement during NGH production when using horizontal wells. COMSOL Multiphysics is a simulation software that enables the formulation of partial differential equations to model hydrate formation and decomposition [36]. But the numerical instability issues may arise in long-term hydrate production simulations.

In this study, we investigate the settlement characteristics of the NGH reservoir with heterogeneous hydrate distribution during gas production via horizontal well depressurization. A fully coupled thermal-hydraulic-mechanical-chemical (THMC) numerical simulation approach is employed to analyze the process. Additionally, the effects of gas production pressure and well location on reservoir settlement behavior are systematically examined.

2 Numerical Simulation Method

In this study, the hydrate reservoir is constructed based on the two-phase flow in porous media. We enhance the chemical processes and engineering thermophysical processes to describe dissociation of the hydrate reservoir. The reliability of the model has been verified in our previous paper [25].

2.1 Continuity Equations for Two Phases Flow

The Continuity Equations of gas and water flow in porous media can be written as:

$$\rho_w C_{p,w} \frac{\partial P_w}{\partial t} + \nabla \rho_w v_w = \dot{m}_w + \rho_w C_{p,w} \frac{\partial P_g}{\partial t} \quad (1)$$

$$\rho_g S \frac{\partial P_g}{\partial t} + \nabla \rho_g v_g = \dot{m}_g + \rho_g C_{p,w} \frac{\partial P_w}{\partial t} \quad (2)$$

$$v_i = -\frac{K_h k_{r,i}}{\mu_i} (\nabla P_i + \rho_i g \nabla D), i = g, w \quad (3)$$

where, $i = g, w$ means non-wetting and wetting, respectively, ρ_i is density, v_i is Darcy's velocity, μ_i is dynamic viscosity, g is the gravitational acceleration, P_i is fluid pressure, $C_{p,w}$ is a function related to pore pressure and saturation and can be defined as:

$$C_{p,w} = -n_{wg} \frac{\partial S_w}{\partial P_c} \quad (4)$$

$$n_{wg} = n(1 - S_h) - n_r \quad (5)$$

$$P_c = P_g - P_w = P_0 (S_w^{-1/m} - 1)^{(1-m)} \quad (6)$$

where, n and n_{wg} are initial porosity and effective porosity, respectively, n_r is residual volume fraction, $n_r = 0.248$ [37]; P_c is pore pressure, P_g and P_w are the pressure of the non-wetting phase and wetting phase, respectively, S_w and S_h are wetting phase saturation and hydrate saturation.

In Eq. (7), S is the storage coefficient which can be defined as:

$$S = n_{wg} S_g \chi_f + C_{p,w} \quad (7)$$

$$\chi_f = \frac{1}{\rho_g} \frac{\partial \rho_g}{\partial P_g} \quad (8)$$

where, χ_f is the gas compression coefficient. According to the gas state equation, the relationship between the density of methane gas, temperature, and pressure can be expressed as:

$$\rho_g = \frac{P_g M_g}{R T} \left[1 + \frac{9}{128} \frac{P_g}{P_{g,c}} \frac{T_c}{T} \left(1 - 6 \frac{T_c^2}{T^2} \right) \right]^{-1} \quad (9)$$

$$\frac{\partial \rho_g}{\partial P_g} = \frac{M_g}{RT} \left[1 + \frac{9}{128} \frac{P_g}{P_{g,c}} \frac{T_c}{T} \left(1 - 6 \frac{T_c^2}{T^2} \right) \right]^{-2} \quad (10)$$

where, $P_{g,c}$ and T_c are the critical pressure and critical temperature, and the corresponding values are 4640 kPa and 190.8 K, respectively.

The coefficient K_h and $k_{r,i}$ in Eq. (11) are absolute permeability and relative permeability, respectively. Based on the Amyx empirical formulation [38,39], the expression of the absolute permeability is as follows:

$$K = \begin{cases} K_i \times (n_{wg})^{0.86}, & n_{wg} < 0.11 \\ K_f \times 10^8 \times (n_{wg})^{9.147}, & n_{wg} \geq 0.11 \end{cases} \quad (11)$$

The relative permeability of fluid and gas was calculated by Formulas (12) and (13).

$$k_{rw} = S_w^L [1 - (1 - S_w^{1/m})^m]^2 \quad (12)$$

$$k_{rg} = (1 - S_w)^L (1 - S_w^{1/m})^{2m} \quad (13)$$

where m and L are coefficient taken as 0.5 and 0.7, respectively [38,39].

2.2 Kinetic Model of Hydrate Decomposition

NGH is composed of water molecules and methane molecules. The NGH decomposition and synthesis reaction is as follows:



Based on the Kim-Bishnoi empirical formulation [40], the production rate of methane gas is calculated by Formula (16):

$$\dot{m} = k_d M_g A_s (P_e - P_g) \quad (15)$$

where, M_g is the molar mass of methane gas, P_e is the three-phase equilibrium pressure of NGH, $P_e = 1.15 \exp(49.3185 - \frac{9459}{T})$, k_d is the decomposition constant of NGH, which is described by the following empirical formula:

$$k_d = k_0 e^{-\frac{\Delta E_a}{RT}} \quad (16)$$

where, k_0 is taken as $3.6 \times 10^4 \text{ mol/(m}^2 \cdot \text{Pa} \cdot \text{s)}$ [41], R is the ideal gas constant, T is temperature, ΔE_a is the kinetic energy, J , $\Delta E_a/R = 9753.73K$ [42], A_s in Eq. (18) is the reaction area of NGH particles [43], which is described by the following empirical formula:

$$A_s = n S_h A_{geo} \quad (17)$$

where, A_{geo} is the surface area ratio of hydrate particles per unit volume, assuming that the natural gas hydrate particles are spherical particles, the calculation is $7.5 \times 10^5 \text{ m}^{-1}$ [44].

2.3 Energy Conservation Equation

The decomposition of natural gas hydrate sediments is a process of absorbing heat from the surrounding environment, and the heat exchange process can be controlled by heat conduction and heat convection equations.

$$(\rho C)_{eq} \frac{\partial T}{\partial t} + (\rho_g C_g v_g + \rho_w C_w v_w) \cdot \nabla T - \nabla \cdot (\lambda_{eq} \nabla T) = Q \quad (18)$$

$$(\rho C)_{eq} = n \rho_g S_g C_g + n \rho_w S_w C_w + n \rho_h S_h C_h + (1 - n) \rho_s C_s \quad (19)$$

$$\lambda_{eq} = n S_g \lambda_g + n S_w \lambda_w + n S_h \lambda_h + (1 - n) \rho_s \lambda_s \quad (20)$$

$$Q = -\dot{m}_h \Delta H_D - n \rho S_g \sigma_g \frac{\partial P_g}{\partial t} - \rho_g v_g \sigma_g \nabla P_g \quad (21)$$

$$\Delta H_D = c_1 + c_2 T \quad (22)$$

where, T represents the temperature, which is the dependent variable of the energy governing equation. C_g , C_w , C_h , C_s are the specific heat capacities of methane non-wetting phase, wetting phase, NGH, and skeleton particles, respectively. λ_g , λ_w , λ_h , λ_s are thermal conductivity of methane non-wetting phase, wetting phase, NGH, and skeleton particles, respectively. ΔH_D is the enthalpy of the methane decomposition process. v_g , v_w are Darcy velocities of gas and water. σ_g is the Joule–Thomson throttling coefficient [44,45] (Appendix A).

2.4 Mechanical Models

We have assumed that the solid particles cannot be compressed, and the settlement of formation only causes the compaction of pore volume. Then the real porosity of the formation can be expressed as:

$$n_p = 1 - (1 - n_0)/J \quad (23)$$

$$J = \det(F) \quad (24)$$

$$F = I + \left[\frac{\partial u}{\partial x} \right] \quad (25)$$

where n_p and n_0 are the real and initial porosities, J is the determinant of the deformation gradient F , I is identity tensor. u represents the dependent variable of the mechanical equilibrium equation, and the main governing equations are described as follows [46–48]:

$$\rho_c \frac{\partial^2 u}{\partial t^2} = \nabla \cdot S + F \cdot v \quad (26)$$

where, ρ_c is the equivalent density of hydrate-bearing sediment, and its calculation formula is $\rho_c = (1 - n)\rho_s + nS_h\rho_h$, n is the porosity. ρ_s , ρ_h are density of skeleton particles and hydrate, respectively. σ is the stress tensor which can be expressed as:

$$\sigma = \sigma_0 + C : \varepsilon \quad (27)$$

where, σ_0 is the initial stress tensor. C is the elastic coefficient matrix. ε is the total strain tensor, which can be expressed by displacement gradient. The calculation formula of displacement gradient is as follows:

$$\varepsilon = \frac{1}{2}(\nabla u + \nabla u^T) \quad (28)$$

According to the incremental theory:

$$\varepsilon = \varepsilon_{ij}^e + \varepsilon_{ij}^p \quad (29)$$

The ε_{ij}^e and ε_{ij}^p are elastic and the plastic strain, respectively.

The plastic strain increment is calculated using the plastic potential energy function:

$$\dot{\varepsilon}_{ij}^p = \lambda \frac{\partial Q}{\partial \sigma'} \quad (30)$$

The potential Q is written in terms of at most three invariants of Cauchy's stress tensor. $Q = Q(I_1(\sigma'), J_2(\sigma'), J_3(\sigma'))$, $I_1(\sigma')$, $J_2(\sigma')$, $J_3(\sigma')$ are the invariants of stress tensor. λ is a scalar that is determined from the consistency condition to keep the stress point on the yield surface, $f(\sigma') = 0$.

In this paper, Mohr-Coulomb yield criterion is used to judge the yield conditions:

$$f(\sigma'_1, \sigma'_3, S_h) = \sigma'_1 - \sigma'_3 N_\varphi - 2c'(S_h) \sqrt{N_\varphi} \quad (31)$$

where, σ'_1 and σ'_3 are the maximum and the minimum principal stress, respectively. $N_\varphi = (1 + \sin \varphi') / (1 - \sin \varphi')$; φ' is the internal friction angle; c' is the cohesion.

Mechanical Properties of NGH Bearing Sediment

The relationship between the cohesion (c), internal friction angle (φ), secant modulus (E_i) with hydrate saturation (S_h) have been tested with triaxial compression experiments and reported in our previous paper [25]. The fitting mathematical relationships are shown as follows:

$$E_i = 563.85S_h + 149 \quad (32)$$

$$c = 0.2647e^{4.077S_h} \quad (33)$$

$$\varphi = -103.69S_h^2 + 35.3S_h + 31.72 \quad (34)$$

3 Model Construction

Fig. 1 presents a schematic diagram of the hydrate reservoir in the Shenu area of the South China Sea. The geological system simulated in this study is situated at site SH02, located on the northern slope of Shenu area. NGH deposits were confirmed in this region during the gas hydrate drilling expedition (GMGS6) conducted by the Guangzhou Marine Geological Survey [49]. Logging results indicate that the water depth ranges from 800 m to 1500 m, with NGH deposits found at depths between 207 and 253.4 m below the seafloor (mbsf). The saturation levels of gas hydrate and free gas vary with depth. Within the interval of 207.8–253.4 mbsf, no free gas is observed, while the saturation of gas hydrate ranges from 0%–54.5%, with an average saturation of 31%. Free gas was detected at depths range between 253.4 and 278 mbsf, with an average saturation of 13%.

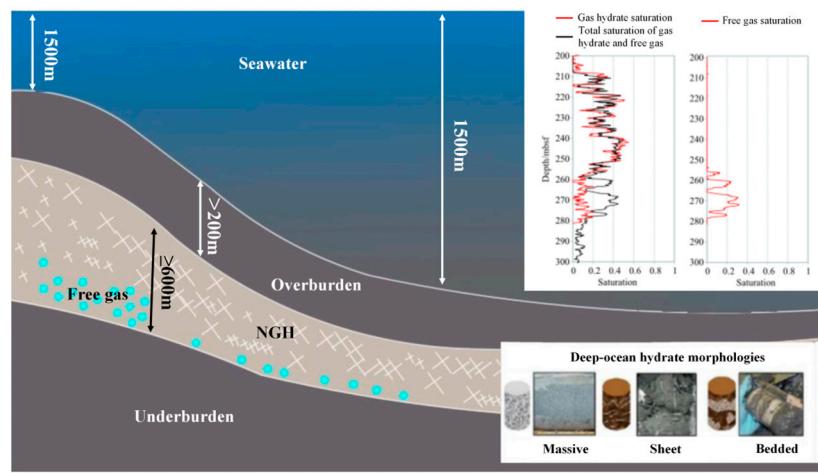


Figure 1: Schematic diagram of hydrate reservoir in Shenu area, South China Sea [50].

3.1 Domain Discretization and Boundary Condition

Fig. 2 shows mapped mesh for hydrate bearing sediment, boundary condition and production well location schematic built in COMSOL. The geometric model is simplified to a two-dimensional model without considering the rate of pressure decay along the horizontal wellbore during production. The model is 73 m in the Y direction, ± 25 m in the X direction, and 1 m in thickness, respectively. The diameter of horizontal wellbore is 0.16 m. The total number of grids are discretized into 32,488 elements. Considering phase change, mass transportation and other complex condition around the wellbore, we chose the triangular grid structure while increasing the mesh grid density near the borehole.

Since the wellhead is far from the model boundary, a constant pressure and temperature boundary with initial condition is used. And a constant production pressure is set to simulate the depressurization production process. To prevent reformation of NGH and ice formation, the temperature of the production well is set as constant value equal to initial temperature. Loading pressure of 12.07 MPa is applied to the upper boundary. TOP1 point located at the top boundary of model is set to monitor the reservoir settlement. The lower boundary is subject to a fixed constraint, while left and right boundaries are subject to roller constraints. Other parameters employed in our model are listed in Table 1.

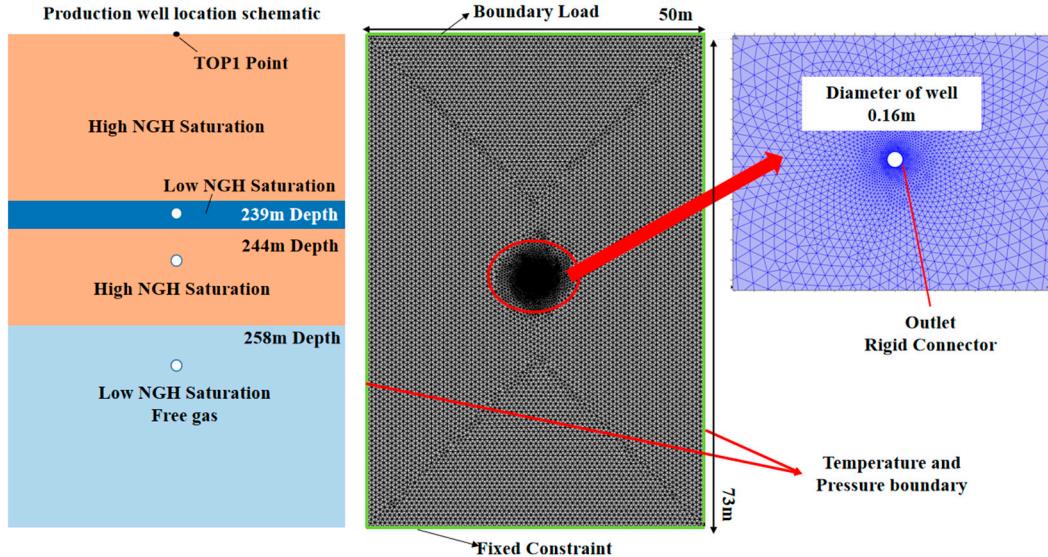


Figure 2: Mapped mesh of the NGH reservoir, boundary condition and production well location setting in the simulation model.

Since the wellhead is far from the model boundary, a constant pressure and temperature boundary with initial condition is used. And a constant production pressure is set to simulate the depressurization production process. To prevent reformation of NGH and ice formation, the temperature of the production well is set as constant value equal to initial temperature. Loading pressure of 12.07 MPa is applied to the upper boundary. TOP1 point located at the top boundary of model is set to monitor the reservoir settlement. The lower boundary is subject to a fixed constraint, while left and right boundaries are subject to roller constraints. Other parameters employed in our model are listed in Table 1.

$$P_{x=\pm 25} = P_{initial}, P_{y=-208} = P_{initial}, P_{y=-281} = P_{initial}, P_{well} = P_{out} \quad (35)$$

$$T_{y=-208} = T_{initial}, T_{y=-281} = T_{initial}, T_{x=\pm 25} = T_{initial}, T_{outlet} = T_{initial} \quad (36)$$

$$F_{y=-208} = -P_{initial} \vec{n}, u_{y=-281} = 0, u_{x=\pm 25} \cdot \vec{n} = 0, u_{outlet} = u_{initial} \quad (37)$$

Table 1: Initial conditions, production pressure, and well locations.

Model Name	Run1	Run2	Run3	Run4	Run5	Run6	Run7
production pressure	4.5 MPa	5.0 MPa	5.5 MPa	6.0 MPa	6.5 MPa		
Well location (Z direction)	-244 m	239 m	258 m				

3.2 Initial Conditions and Simulation Parameters

Prior to production of NGH, it is essential to establish the equilibrium pressure-temperature initial conditions to ensure that the simulation model's temperature, pressure and saturation align with the actual geological conditions. Well logging results from the SH02 site indicate that the NGH deposits are located at a

depth of 208 m below sea floor (mbsf) in a water column of 800 m. Due to the absence of *in-situ* temperature measurements, the following empirical formula is used to calculate the initial formation temperature.

$$T = T_0 + \text{grad} \times Z \times 10^{-3} \quad (38)$$

$$T = T_0 + \text{grad} \times Z \times 10^{-3} \quad (39)$$

where, T_0 is the sea floor temperature of 275.15 K, grad is geothermal gradient of 44 K/km, Z is depth of formation in meters ranging from 208 m to 281 m. Then the pore water pressure can be calculated using the hydrate pressure-temperature equilibrium curve. Initial temperature and pressure of the NGH reservoir are shown in Fig. 3. Other initial parameters used in this study are listed in Table 2. And the distribution of NGH saturation is set based on the log data obtained from GMGS6-SH02, as shown in Fig. 3c.

Table 2: Simulation parameters [25,44,45].

Parameters	Value	Parameters	Value
Absolute porosity/n	0.47	Viscosity of water/ μ_w	1e-3 Pa·s
specific heat of NGH/ C_h	2170 J/(kg·K)	Thermal conductivity of NGH/ λ_h	0.7 W/(m·K)
specific heat of water/ C_w	4186 J/(kg·K)	Thermal conductivity of CH_4/λ_g	0.033 W/(m·K)
specific heat of CH_4/C_g	2191 J/(kg·K)	Thermal conductivity of water/ λ_w	0.6 W/(m·K)
specific heat of sand/ C_s	800 J/(kg·K)	Thermal conductivity of sand/ λ_s	1.35 W/(m·K)
Joule-Thomson coefficient/ σ_g	-1.5e-4 J/(kg·Pa)	Density of NGH/ ρ_h	913 kg/m ³
Hydration number	5.75	Density of water/ ρ_w	1000 kg/m ³
Rate constant of dissociation/ k_0	3.6e4 mol/(m ² ·Pa·s)	Density of sand/ ρ_s	2000 kg/m ³
Viscosity of CH_4/μ_g	1.81e-5 Pa·s	residual volume fraction/ n_r	0.248

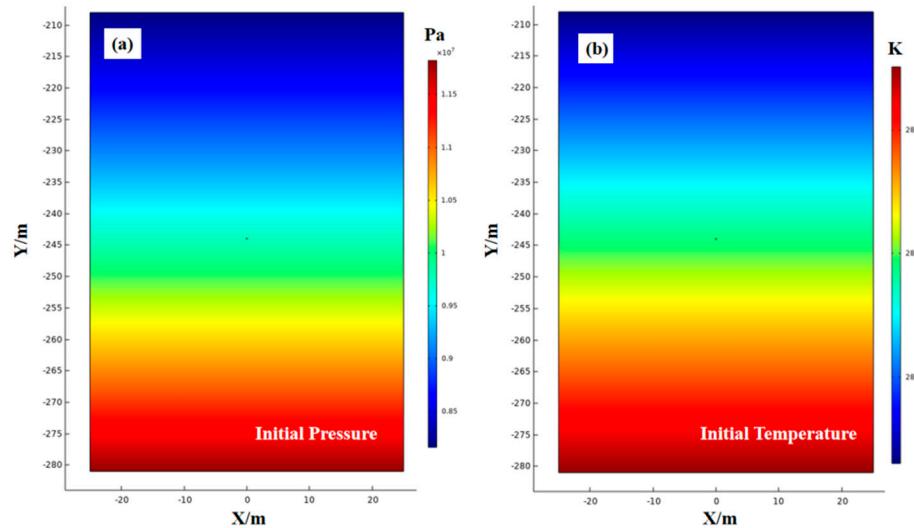


Figure 3: Cont.

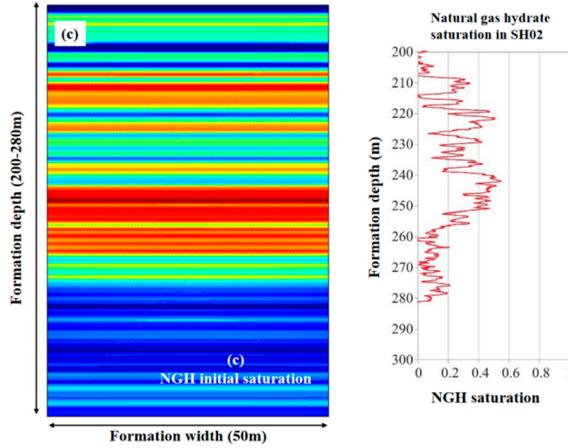


Figure 3: Initial pressure (a), temperature (b), NGH saturation (c), and logging data of SH02 [49].

4 Simulation Results and Discussion

4.1 Effect of Outlet Pressure on Gas Production Process

Fig. 4 shows the pore pressure and temperature change with production time. A significant pressure drop is observed nearby the production well in the early stage of production, and then it remains in a relatively stable state. The pressure drop area shows as a funnel shape, and its decomposition range is closely related to the production pressure; specially, a lower production pressure results in a larger decomposition area. In general, due to the low permeability of the reservoir, the decomposition area from the pressure drop remains limited even after six months of production. It is well known that the production pressure is the main factor to control the hydrate decomposition rate in depressurization method. After half a year of production, the radius of decomposition region is about 2 m with production pressure of 6.5 MPa and it is 4 m with production pressure of 4.5 MPa. Since hydrate decomposition and free gas production is an endothermic process, a decrease in production pressure accelerates both hydrate decomposition and gas production, leading to a more pronounced temperature drop near the wellbore.

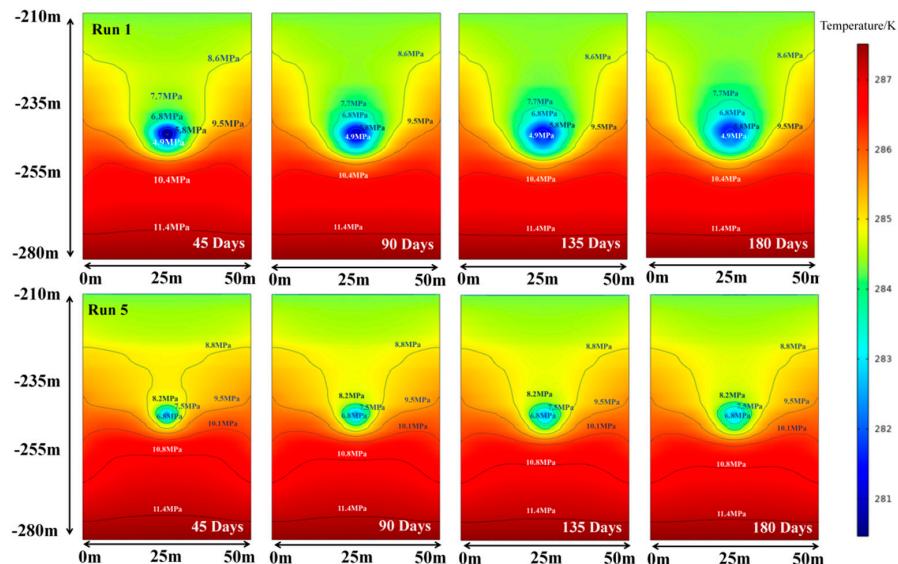


Figure 4: Formation pore pressure and temperature change with production time.

Fig. 5 shows the gas and water production rate for per unit well length with different production pressure. Since there is free gas in the NGH reservoir, and no drainage depressurization process in the numerical simulation, water and gas can be produced instantly at the beginning of production. It shows a high initial gas production rate, then rapid decayed for the consumption of free gas. During the initial production phase, the migration of free gas and water near the wellbore towards the wellbore leads to an enhanced production rate; however, subsequent migration of free gas and water from low permeability formations is hindered, resulting in a rapid decline in production rate [51]. As the production progresses, the system pressure gradually decreases due to the discharge of free gas and water. Once it reaches the equilibrium pressure of natural gas hydrates (NGH), NGH decomposition occurs to supplement the gas source, resulting in enhancement in production rate. It has been reported that the average gas production rate of the second hydrate production test in the Shenhua area of the South China Sea is 2.9×10^4 m³/day [49]. Under similar conditions with a horizontal well length, simulation results indicate a production rate of 3.8×10^4 m³/day at an outlet pressure of 6.5 MPa, which aligns closely with field data. As the production pressure is reduced to 4.5 MPa, the gas and water production rates increase to 2 times and the produced cumulative gas increase to 3 times. Therefore, reducing production pressure during the production can improve gas production effectively.

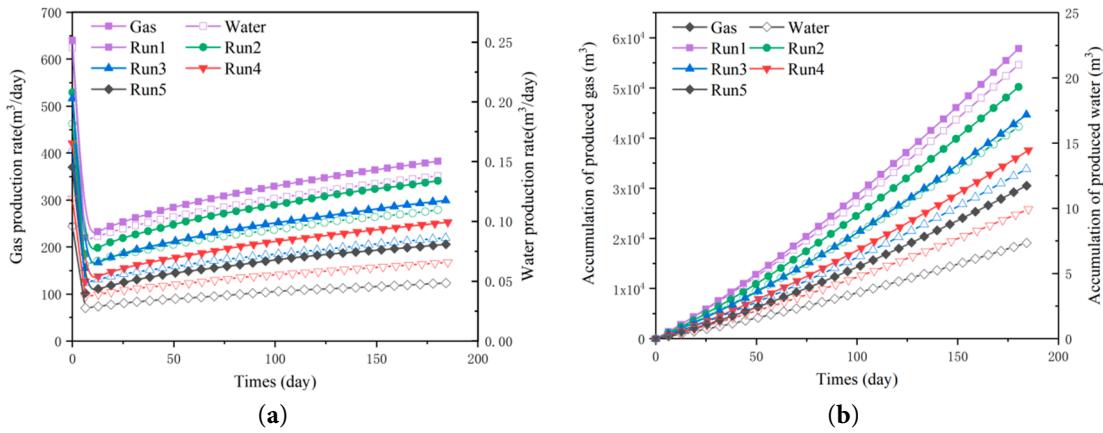


Figure 5: Gas and water production rate (a), accumulation of produced gas and water (b) 1 m well length with different production pressure.

Fig. 6 shows NGH saturation change with production time. Since the NGH saturation reservoir is stratified, the degree of hydrate saturation within the pore increases inversely with respect to the permeability, the permeability is diminished as a result of the presence of NGH in the pore. The production pressure drop can be transmitted quickly into depth of the low S_h area, which shows a significant ahead of NGH decomposition front in low S_h layers than that in high S_h region. Since production pressure is the main factor to control NGH decomposition rate in the production process, the radius of decomposition region is about 2 m with production pressure of 6.5 MPa after half a year of production, and it is 6 m with production pressure of 4.5 MPa. Additionally, there are a few secondary hydrates occurring near the hydrate dissociation front. The heat absorption associated with NGH dissociation alters the pressure-temperature conditions of phase equilibrium, leading to the reformation of NGH.

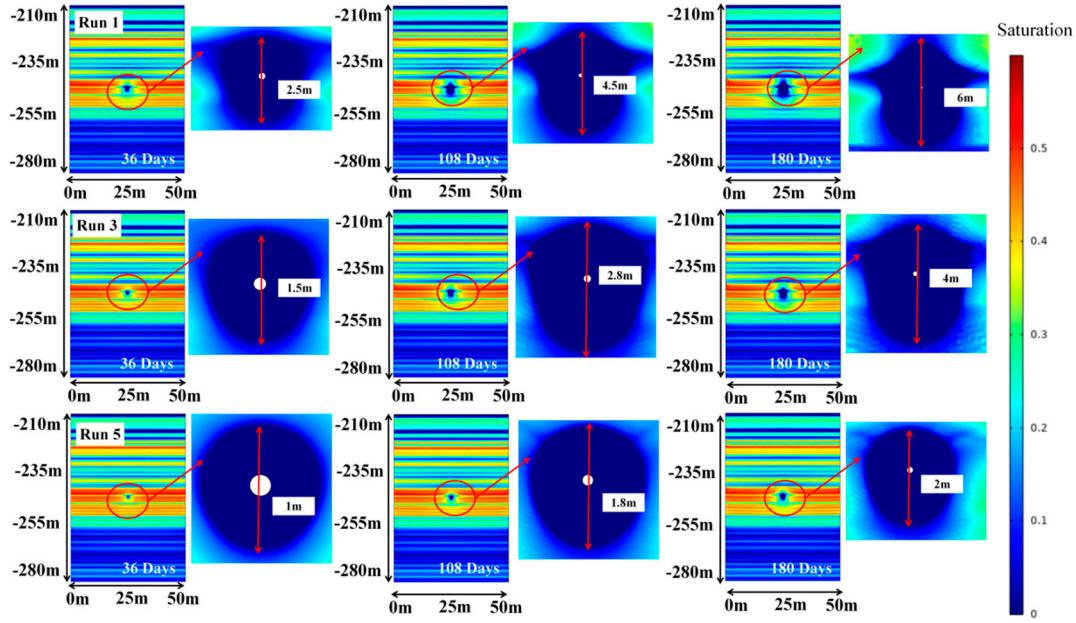


Figure 6: NGH saturation variation with production time and different production pressure.

Fig. 7 shows the reservoir settlement characters with different production pressure. The primary settlement is concentrated in the upper deposit around the wellbore, with wellbore settlement values of 0.032 m and 0.018 m observed at pressures of 4.5 MPa and 6.5 MPa, respectively. Meanwhile, the settlement of TOP1 point in Run1 reaches to nearly 0.1 m with production pressure of 4.5 MPa. And the settlement curves show that settlement displacement will continue to increase with the NGH production. The top boundary exhibits a U-shaped sag over a production period of 180 days. The settlement rate of formation refers to subsidence in per unit time. Its maximum value occurs at the beginning of production and decreasing with production time, which is very similar to the gas production rate curve. Lower production pressures induce greater settlement and sag in the top boundary, increasing likelihood of effective stress concentration and geological landslide. This is corroborate by the settlement rate observed at the upper boundary. With the free gas and water discharged, the mechanical weakness of reservoir caused by NGH decomposition become the main reason for reservoir settlement. And the settlement rate shows a gradual decline with the production time.

4.2 Influence of Horizontal Well Location

Fig. 8 shows NGH saturation distribution contours with different well location.

The NGH saturation exerts a significant influence on the permeability, induces regional distribution of hydrate decomposition. The decomposition shape of the Run6 model starts with the wellbore and diffuses along the low saturation hydrate layer. The decomposition area forms a diamond-shaped distributed after approximately 180 days, and the movement of the decomposition interface in the lower layer of the well site is slower due to higher NGH saturation of formation. The decomposition interface of the Run3 exhibits an approximate circular shape with a radius ranging from 4 m after a period of 180 days. The occurrence of uniformly distributed, highly saturated hydrate layers surrounding the wellbore is responsible for this phenomenon. The well location of the Run7 model is located within the free gas layer, thereby facilitating the most rapid decomposition of hydrates and resulting in a largest decomposition area.

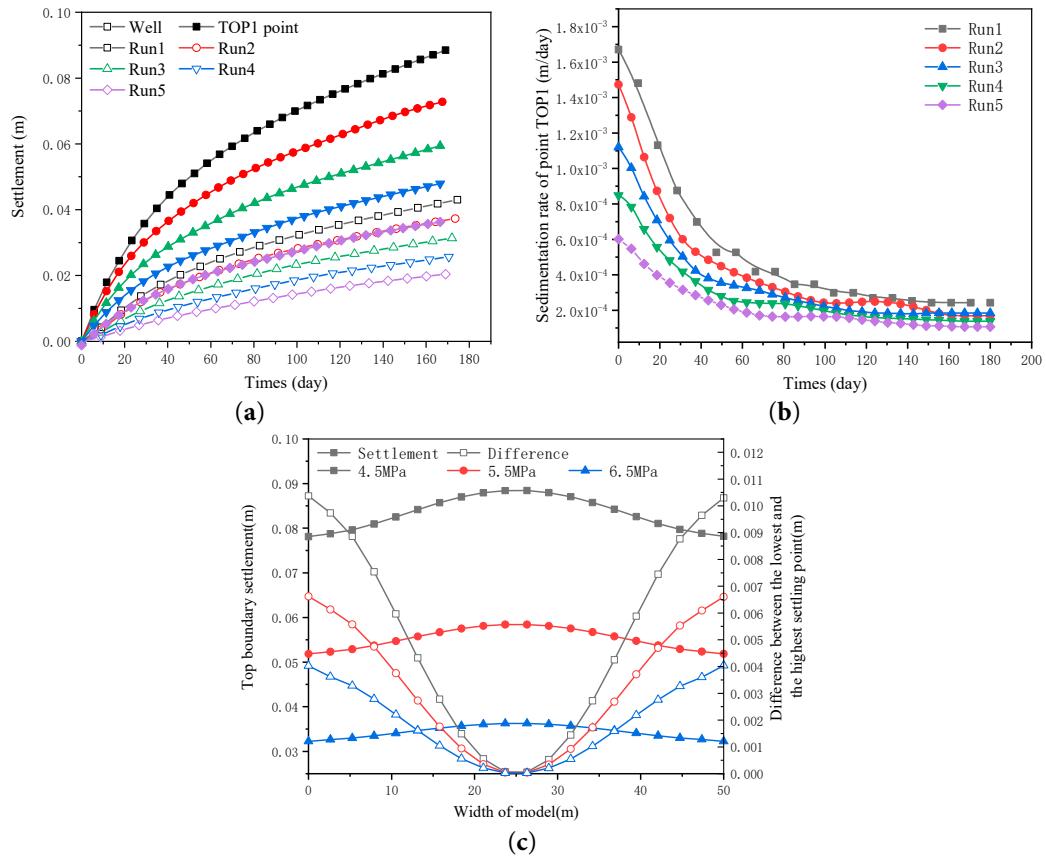


Figure 7: Reservoir settlement characteristics with different production pressure. (a) Settlement, (b) Sedimentation rate (c) Top boundary settlement.

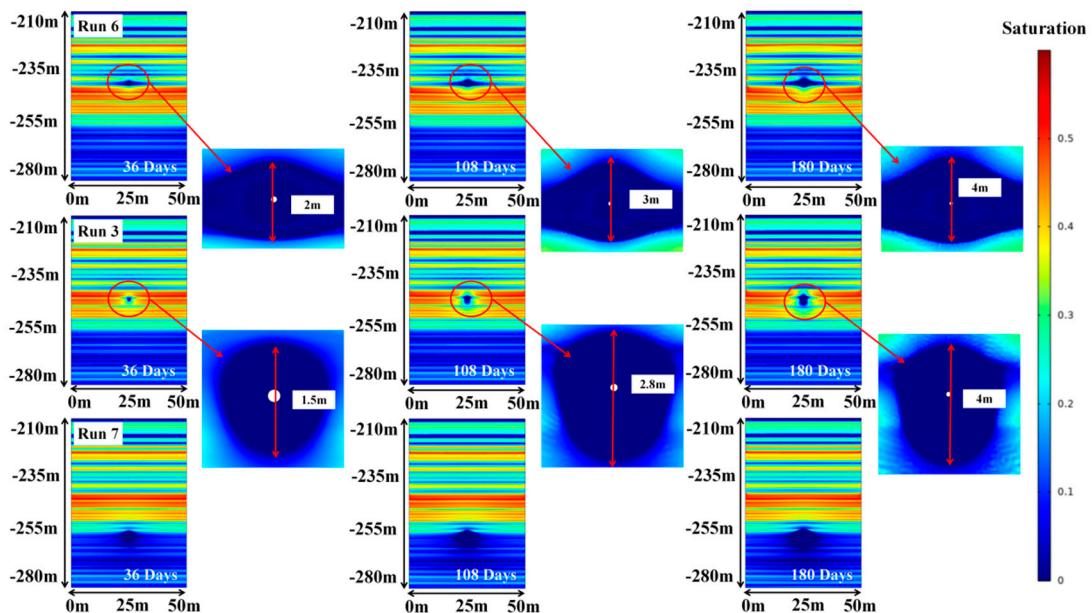


Figure 8: NGH saturation distribution contours with different well location.

Fig. 9 illustrates gas and water production rates and accumulated production for different well location with a well length of 1 m. Initially, production rates are high at the onset of production, rapidly decreasing to a lower value, then gradually rising and stabilizing. During the decline stage of production rate, Run3 exhibits the lowest rate while Run7 is the highest. At the production enhancement stage, Run3's rate gradually aligns with that of Run6, whereas Run7's rate remains relatively constant. The wellbore of Run3, located in a high-saturation hydrate formation, impedes free gas and water migration due to elevated saturation. However, the wellbores of Run6 and Run7, located in lower-saturation hydrate and free gas zone, respectively, indicates the greater permeability, facilitating fluid migration to the wellbores. In the mid-to-late production stages, gas from hydrate dissociation sustains decomposition rate, resulting in a higher recovery rate for Run3 compared to Run6 in later stages. The cumulative gas production of Run7 is highest, as it includes free gas, hydrate-derived gas and dissolved gas from pore liquid.

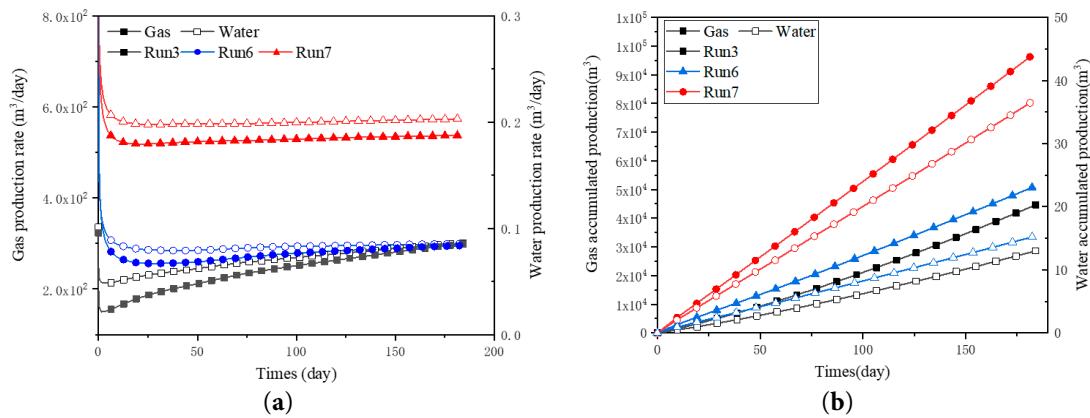


Figure 9: Gas and water production rate (a) and accumulated production (b) with different well location for 1 m well length.

Fig. 10 shows reservoir settlement characteristics under different well locations. It is shown that after 30 days production, the settlement and settlement rate of TOP1 point in the Run3 surpasses that of the Run6. The production well in Run6 is located in the low S_h layer where with relatively high permeability, and the pore pressure drops significantly quicker than Run3, then leading to a faster settlement in Run6. With the continuous gas production, the hydrate decomposition within the pressure drop influence range becomes the main gas source. The pressure drop influence range in Run6 is significantly larger than that in Run3, and the pressure drop gradient at local points is smaller than that in Run3, resulting in a smaller amount of hydrate decomposition at local points in Run6 and a smaller local settlement for formation weakening caused by hydrate decomposition. The settlement of Run7 is the largest among all cases after 180 days gas production. Since the pore pressure drop gradient is more evenly distributed due to the wellbore located in the free gas region in Run7, it shows a minimum sag in its top boundary, which is benefit to prevent the formation of tension fractures caused by uneven settlement. The settlement of Run6 is the smallest among all, while exhibiting the largest sag in its top boundary. And the settlement of Run3 is comparatively lower than that of Run6 due to its location in a highly saturated formation with enhanced strength. Above results show that well location would significantly affect the settlement and gas characteristics of the NGH reservoir. Although locating the production well in the overlying free gas layer would lead to a largest settlement, the formation uneven settlement is minimal in this way, which can effectively reduce the tensile fractures development caused by the uneven settlement. Meanwhile, the gas production rate of this wellbore arrangement can significantly increase the gas production rate.

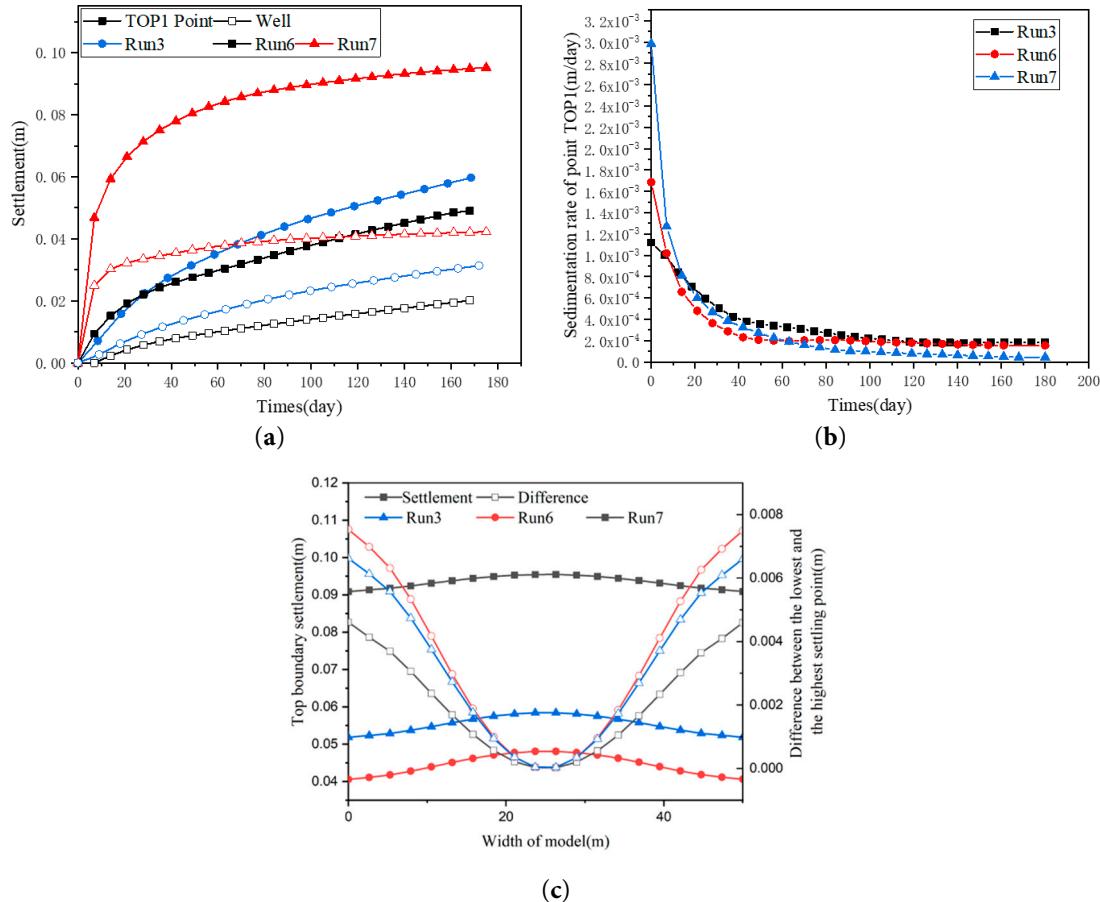


Figure 10: Reservoir settlement characteristics under different well location (a) Displacement of point TOP1, (b) Sedimentation rate of point TOP1, (c) Displacement of upper boundary.

5 Conclusion

This study investigates the settlement behavior of horizontal wellbores in natural gas hydrate (NGH) reservoirs with heterogeneous hydrate distributions. The influences of well placement under various production pressures and single-well production conditions on both gas production and formation settlement are analyzed. Results show that although lower production pressures enhance productivity, but also lead to increased formation settlement and a higher risk of tensile fractures. Previous studies have also indicated that excessively low production pressures may promote secondary hydrate formation, thereby impeding gas flow [52]. Thus, it is crucial to optimize production pressure based on specific NGH reservoir characteristics. Adjusting the horizontal well position can help compensate for the production deficiency caused by high outlet pressure. The optimal configuration appears to place the well near free gas zones within low-saturation hydrate formations. This strategy not only improves gas production but also mitigates uneven settlement, thereby reducing the potential for tensile fracture development.

Acknowledgement: None.

Funding Statement: This work is financially supported by the State Key Research Development Program of China (Grant No. 2021YFC2800905-02), the National Natural Science Foundation of China (Grant No. 52304208), which are gratefully acknowledged.

Author Contributions: Lijia Li: Conceptualization, Methodology, Investigation, Writing—Original Draft. Shu Liu: Software, Formal analysis, Data Curation. Xiaoliang Huang: Writing Review and Editing. Zhilin Qi: Resources, Validation, Investigation, Writing Review. All authors reviewed approved the final version of the manuscript.

Availability of Data and Materials: The datasets used and analyzed during the current study are available from the corresponding author on reasonable request.

Ethics Approval: Not applicable.

Conflicts of Interest: The authors declare no conflicts of interest.

Nomenclature

A_s	Dissociation surface area of hydrate particles of unit volume
$C_{p,w}$	Capillary capacity of water
P_g	Pore pressure of methane gas
P_0	Entry pressure
$P_{g,c}$	Critical pressure
n_0	Initial porosity
C_i	specific heat, $i = h, w, g, s$
χ_f	Gas compression coefficient
K_h	Absolute permeability
\dot{m}_g	Production rate of methane gas
σ_g	Joule-Thomson coefficient
N_H	Hydration number
k_0	Rate constant of dissociation
μ_i	Dynamic viscosity, $i = g, w$
T	Temperature
σ'	Effective stress
σ	Stress tensor
φ	Internal friction angle
P_{initial}	Initial Pressure
P_{out}	Outlet pressure
T_{outlet}	Outlet temeprature
u_{initial}	Initial settlement
A_{geo}	Surface areas to volume ratio of hydrate particles
P_c	Capillary tube pressure
P_w	Pore pressure of water
P_e	Gas pressure of three-phase equilibrium state of hydrate
n_{wg}	Effective porosity
n	Porosity
ε_v	Volumetric strain of sediment
λ_i	Thermal conductivity, $i = h, g, w, s$
$k_{r,i}$	Relative permeability, $i = g, w$
ΔH_D	Enthalpy of the methane decomposition process
ρ_i	Density, $i = h, w, s$
ΔE_a	Kinetic energy
k_d	Decomposition constant of NGH
S_i	Saturation, $i = h, g, w$
T_c	Critical temperature
v_i	Darcy's velocity tensor, $i = g, w$
ε	Total strain tensor
c	Cohesion

P_{well}	Pressure of well
T_{initial}	Initial temperature
u_{outlet}	Settlement of well
F	Formation weight stress

Appendix A

The energy equation is written in the form of enthalpy and temperature.

$$\frac{\partial}{\partial t} [\varphi(\rho_g S_g H_g + \rho_w S_w H_w + \rho_h S_h H_h) + (1 - \varphi)\rho_s H_s] + \nabla(\rho_g S_g H_g + \rho_w S_w H_w) = \nabla(\lambda \nabla T) \quad (\text{A1})$$

where, H_i , $i = g, w, h$ is total enthalpy per unit volume of methane gas, water, and hydrate, respectively, $C_i = \frac{\partial H_i}{\partial T}$, $i = g, w, h$, is the specific heat capacity of methane gas, water and hydrate, respectively.

$$dH_i = \frac{\partial H_i}{\partial T} dT + \frac{\partial H_i}{\partial p} dp = C_i dT + \sigma_i dp \quad (\text{A2})$$

For gas, need to consider the second term of the right side of Eq. (A2), that is, Joule-Thomson throating coefficient, $\sigma g = \frac{\partial H_g}{\partial p}$, $\sigma_g = -1.5 \times 10^{-4} \text{ J/kg} \cdot \text{Pa}$.

Considering (A2), the first term of the left side of Eq. (A1) gets:

$$\frac{\partial}{\partial t} (\varphi \rho_g S_g H_g) = \varphi \left[\frac{\partial}{\partial t} (\rho_g S_g) H_g + \rho_g S_g \frac{\partial}{\partial t} H_g \right] \quad (\text{A3})$$

$$\frac{\partial}{\partial t} (\varphi \rho_w S_w H_w) = \varphi \left[\frac{\partial}{\partial t} (\rho_w S_w) H_w + \rho_w S_w C_w \frac{\partial}{\partial t} T \right] \quad (\text{A4})$$

$$\frac{\partial}{\partial t} (\varphi \rho_h S_h H_h) = \varphi \left[\frac{\partial}{\partial t} (\rho_h S_h) H_h + \rho_h S_h C_h \frac{\partial}{\partial t} T \right] \quad (\text{A5})$$

$$\frac{\partial}{\partial t} (1 - \varphi) \rho_s H_s = (1 - \varphi) \rho_s C_s \frac{\partial}{\partial t} T \quad (\text{A6})$$

Then, the second term of the left side of Eq. (A1) can be written as:

$$\nabla(\rho_g v_g + \rho_w v_w H_w) = \nabla(\rho_g v_g) H_g + \rho_g v_g \nabla H_g + \nabla(\rho_w v_w) H_w + \rho_w v_w \quad (\text{A7})$$

Add (A3), (A7), substituting into (A1), gets

$$\begin{aligned} & [\varphi \frac{\partial}{\partial t} (\rho_g S_g) + \nabla(\rho_g v_g)] H_g + [\varphi \frac{\partial}{\partial t} (\rho_w S_w) + \nabla(\rho_w v_w)] H_w + [\varphi \frac{\partial}{\partial t} (\rho_h S_h) + \nabla(\rho_h v_h)] H_h + \\ & [\varphi(\rho_g S_g C_g + \rho_w S_w C_w + \rho_h S_h C_h) + (1 - \varphi)\rho_s C_s] \frac{\partial T}{\partial t} + \varphi \rho_g S_g \sigma_g \frac{\partial p_g}{\partial t} + (\rho_g v_g C_g + \rho_w v_w C_w) \nabla T \\ & + \rho_g v_g \sigma_g \nabla p_g = \nabla(\lambda \nabla T) \end{aligned} \quad (\text{A8})$$

Considering Eq. (A2), (A8) can be written as:

$$m_g H_g + m_w H_w - m_h H_h + (\rho C)_{eq} \frac{\partial T}{\partial t} + (\rho_g v_g C_g + \rho_w v_w C_w) \nabla T + \nabla(\lambda \nabla T) = -\varphi \rho_g S_g \sigma_g \frac{\partial p_g}{\partial t} - \rho_g v_g \sigma_g \nabla p_g \quad (\text{A9})$$

where,

$$(\rho C)_{eq} = \varphi(\rho_g S_g C_g + \rho_w S_w C_w + \rho_h S_h C_h) + (1 - \varphi)\rho_s C_s \quad (A10)$$

Considering the energy balance conditions for hydrate dissociation:

$$m_h \nabla H = m_g H_g + m_w H_w - m_h H_h \quad (A11)$$

where ∇H is the latent heat for hydrate dissociation.

$$(\rho C)_{eq} \frac{\partial T}{\partial t} + (\rho_g v_g C_g + \rho_w v_w C_w) \nabla T + \nabla(\lambda \nabla T) = Q \quad (A12)$$

where,

$$Q = -m_h \nabla H - \varphi \rho_g S_g \sigma_g \frac{\partial p_g}{\partial t} - \rho_g v_g \sigma_g \nabla p_g \quad (A13)$$

References

- Ul HI, Qasim A, Lal B, Zaini DB. Mini review on environmental issues concerning conventional gas hydrate inhibitors. *Process Saf Prog*. 2022;41(S1):S129–34. [\[CrossRef\]](#).
- Li YX, Zhang ZB, Li SD. Enhanced strategies for depressurization in offshore natural gas hydrate exploitation: An in-depth investigation into pathway optimization and production stability mechanisms. *Mar Pet Geol*. 2024;170:107107. [\[CrossRef\]](#)
- Majumdar D, Gupta M, Shastri Y, Mahajani SM. Environmental and economic assessment of gas production from gas hydrate reserves in Krishna-Godavari basin in the Indian offshore. *Sustain Energy Technol Assess*. 2023;56:103050. [\[CrossRef\]](#).
- Gaidukova O, Misura S, Strizhak P. Key Areas of Gas Hydrates Study: Review. *Energies*. 2022;15(5):1799. [\[CrossRef\]](#).
- Kondori J, Zendehboudi S, Hossain ME. A review on simulation of methane production from gas hydrate reservoirs: Molecular dynamics prospective. *J Pet Sci Eng*. 2017;159:754–72. [\[CrossRef\]](#).
- Lee S, Park S, Lee Y, Seo Y. Thermodynamic and ¹³C NMR spectroscopic verification of methane-carbon dioxide replacement in natural gas hydrate. *Chem Eng J*. 2013;225:636–40. [\[CrossRef\]](#).
- Liang YP, Tan YT, Luo YJ, Zhang YY, Li B. Progress and challenges on gas production from natural gas hydrate-bearing sediment. *J Clean Prod*. 2020;261:121061. [\[CrossRef\]](#).
- Liu WG, Luo TT, Li YH, Song YC, Zhu Y, Liu Y, et al. Experimental study on the mechanical properties of sediments containing CH₄ and CO₂ hydrate mixtures. *J Nat Gas Sci Eng*. 2015;32:20–7. [\[CrossRef\]](#).
- Ye HY, Chen JY, Yao YX, Dong P, Chen DY, Niu MY, et al. How injecting CO₂ enhances marine natural gas hydrate exploitation: Review and prospect. *Mar Pet Geol*. 2024;170:107107. [\[CrossRef\]](#).
- Almenningen S, Gautepllass J, Hauge LP, Barth T, Fernø MA, Ersland G. Measurements of CH₄ and CO₂ relative permeability in hydrate-bearing sandstone. *J Pet Sci Eng*. 2019;177:880–8. [\[CrossRef\]](#).
- An SH, Le VS, Chon BH. Production characteristics of gas hydrate reservoirs associated with steam injection and depressurization rate. *Int J Oil Gas Coal Technol*. 2018;19(3):296–315. [\[CrossRef\]](#).
- Yang X, Sun CY, Yuan Q, Ma PC, Chen GJ. Experimental study on gas production from methane hydrate-bearing sand by hot-water cyclic injection. *Energy Fuels*. 2010;24(11):5912–20. [\[CrossRef\]](#).
- Zhao E, Hou J, Ji Y, Liu Y, Bai Y. Enhancing gas production from Class II hydrate deposits through depressurization combined with low-frequency electric heating under dual horizontal wells. *Energy*. 2021;233:121137. [\[CrossRef\]](#).
- Lee T, Ahn T, Park S. Simulation study of gas hydrate production using various thermal stimulation methods in the Ulleung Basin of the Korean East Sea. *Pet Sci Technol*. 2025;1–20. [\[CrossRef\]](#).
- Li L, Li X, Wang Y, Luo Y, Li B. Analyzing the applicability of *in situ* heating methods in the gas production from natural gas hydrate-bearing sediment with field scale numerical study. *Energy Rep*. 2020;6:3291–302. [\[CrossRef\]](#).

16. Feng JC, Wang Y, Li XS. Entropy generation analysis of hydrate dissociation by depressurization with horizontal well in different scales of hydrate reservoirs. *Energy*. 2017;125:62–71. [\[CrossRef\]](#).
17. Liu LP, Sun ZL, Zhang L, Wu NY, Qin YC. Progress in global gas hydrate development and production as a new energy resource. *Acta Geol Sin Engl*. 2019;93(3):731–55. [\[CrossRef\]](#).
18. Ye JL, Qin XW, Xie WW, Lu H, Ma B, Qiu H, et al. The second natural gas hydrate production test in the South China Sea. *China Geol*. 2020;2:197–209. [\[CrossRef\]](#).
19. Pandey MR, Priest JA, Hayley JL. Influence of test conditions and soil properties on the geomechanical response of hydrate-bearing sands. *Can Geotech J*. 2025;62:0378. [\[CrossRef\]](#).
20. Yan CL, Ren X, Cheng YF, Song BJ, Li Y, Tian WQ. Geomechanical issues in the exploitation of natural gas hydrate. *Gondwana Res*. 2020;81:403–22. [\[CrossRef\]](#).
21. Rydz MB. The effect of hydrate formation on the elastic properties of unconsolidated sediment [dissertation]. Golden, CO, USA: Colorado School of Mines; 2014. p. 127.
22. Chuvilin EM, Bukhanov BA, Grebenkin S, Doroshin VV, Iospa AV. Shear strength of frozen sand with dissociation pore methane hydrate: An experimental study. *Cold Reg Sci Technol*. 2018;153:101–5. [\[CrossRef\]](#).
23. Wang AW, Zhang CY, Luo TT, Ma CZ, Zhao Y, You ZS, et al. Mechanical properties of consolidated water-saturated natural gas hydrate-bearing silty-clayey sediments under undrained shearing conditions. *Int J Hydrogen Energy*. 2025;99:996–1009. [\[CrossRef\]](#).
24. Pandey MR, Priest JA, Hayley JL. The Influence of Particle Size and Hydrate Formation Path on the Geomechanical Behavior of Hydrate Bearing Sands. *Energies*. 2022;15(24):9632. [\[CrossRef\]](#).
25. Li LJ, Li XS, Wang Y, Huang XL, Qi ZL, Chen Z. Study on the settlement characteristics of hydrate bearing sediment caused by gas production with the depressurization method. *Energy Fuels*. 2024;38(7):5810–21. [\[CrossRef\]](#).
26. Yoneda J, Suzuki K, Oshima M, Muraoka M, Jin Y. Empirical evaluation of the strength and deformation characteristics of natural and synthetic gas hydrate-bearing sediments with different ranges of porosity, hydrate saturation, effective stress, and strain rate. *J Vis Exp*. 2024;11:3. [\[CrossRef\]](#).
27. Priest JA, Best AI, Clayton CRI. A laboratory investigation into the seismic velocities of methane gas hydrate-bearing sand. *J Geophys Res*. 2005;110:B04102. [\[CrossRef\]](#).
28. Kim J, Seol Y, Dai S. The coefficient of earth pressure at rest in hydrate-bearing sediments. *Acta Geotech*. 2021;16(9):2729–39. [\[CrossRef\]](#).
29. Gan BC, Li ZD, Zhang HX, Zhang YZ, Huo WX, Li Z, et al. Optimizing hydrate extraction: Balancing stability and production efficiency. *Fuel*. 2025;384:134088. [\[CrossRef\]](#).
30. Li JW, Zhang Y, Di SJ, Lin L, Zhou Y. Research on hydrate-bearing reservoir deformation and wellbore wall stability during natural gas hydrate exploitation. *Geomech Energy Environ*. 2023;34:100458. [\[CrossRef\]](#).
31. Myshakin E, Garapati N, Seol Y, Gai X, Boswell R, Ohtsuki S, et al. Numerical Simulations of Depressurization-Induced Gas Hydrate Reservoir (B1 Sand) Response at the Prudhoe Bay Unit Kuparuk 7-11-12 Pad on the Alaska North Slope. *Energy Fuels*. 2022;36(5):2542–60. [\[CrossRef\]](#).
32. Lele AF, Kuznik F, Rammelberg HU, Schmidt T, Ruck WKL. Thermal decomposition kinetic of salt hydrates for heat storage systems. *Appl Energy*. 2015;154:447–58. [\[CrossRef\]](#).
33. Jiang MJ, Lu YC, Wang HN, Chen YR. Multi-field coupling analysis of mechanical responses in methane hydrate exploitation with a practical numerical approach combining T+H with DEM. *Comput Geotech*. 2024;166:105978. [\[CrossRef\]](#).
34. Zhao XL, Wang ZL, Zhao YZ, Zuo JQ, Li P, Liang W, et al. Coupled thermal-hydrodynamic-mechanical numerical simulation of natural gas hydrate horizontal well depressurization production: method and application in the South China Sea. *Nat Gas Ind B*. 2022;9(6):548–60. [\[CrossRef\]](#).
35. Cheng FB, Sun X, Wu P, Chen ZX, Yu T, Liu WG, et al. A Fully Coupled Thermo-Hydro-Mechanical-Chemical Model for Methane Hydrate Bearing Sediments Considering the Effect of Ice. *J Mar Sci Eng*. 2023;11:766. [\[CrossRef\]](#).
36. Jiang YJ, Ma XZ, Cheng XZ. Numerical simulation on natural gas hydrate depressurization production considering sediment compression effects. *Energy*. 2024;301:131745. [\[CrossRef\]](#).
37. Li YX, Zhang ZB, Li SD, Li X, Lu C. Numerical investigation of the depressurization exploitation scheme of offshore natural gas hydrate: enlightenments for the depressurization amplitude and horizontal well location. *Energy Fuels*. 2023;37(14):10706–20. [\[CrossRef\]](#).

38. Ghezzehei TA, Kneafsey TJ, Su GW. Correspondence of the Gardner and van Genuchten-Mualem relative permeability function parameters. *Water Resour Res.* 2007;43(10):W10417. [\[CrossRef\]](#).
39. Amyx JW, Bass DM, Whiting RL. Petroleum reservoir engineering physical properties. New York, NY, USA: McGraw-Hill; 1960.
40. Kim HC, Bishnoi PR. Kinetics of methane hydrate decomposition. *Chem Eng Sci.* 1987;42:1645–53. [\[CrossRef\]](#).
41. Clarke M, Bishnoi PR. Determination of the activation energy and intrinsic rate constant of methane gas hydrate decomposition. *Can J Chem Eng.* 2001;79(1):143–7. [\[CrossRef\]](#).
42. Kumar A, Maini B, Bishnoi PR, Clarke M, Zatsepina O, Srinivasan S. Experimental determination of permeability in the presence of hydrates and its effect on the dissociation characteristics of gas hydrates in porous media. *J Pet Sci Eng.* 2010;70:114–22. [\[CrossRef\]](#).
43. Deng XF, Feng JW, Pan SW, Wang ZY, Zhang JB, Chen WQ. An improved model for the migration of fluids caused by hydrate dissociation in porous media. *J Pet Sci Eng.* 2020;188:106876. [\[CrossRef\]](#).
44. Soga K, Klar A. Coupled deformation-flow analysis for methane hydrate production by depressurized wells. In: Proceedings of the 3rd International Biot Conference on Poromechanics; 2005 May 24–27; Norman, OK, USA. [\[CrossRef\]](#).
45. Deng XF, Pan SW, Zhang JB, Wang ZY, Jiang ZY. Numerical investigation on abnormally elevated pressure in laboratory-scale porous media caused by depressurized hydrate dissociation. *Fuel.* 2020;271:117679. [\[CrossRef\]](#).
46. Miyazaki K, Masui A, Yamamoto Y, Ogata Y, Aoki K, Yamaguchi T. Investigation of deformation mechanism for methane hydrate sediment based upon mechanical properties in unloading and reloading process under triaxial compression. In: Proceedings of the 8th International Offshore and Polar Engineering Conference; 2009 Jul 6–11; Osaka, Japan.
47. Dhakal S, Gupta I. Gas hydrates reserve characterization using thermo-hydro-mechanical numerical simulation: a case study of green canyon 955, gulf of Mexico. *Energies.* 2023;16(7):3275. [\[CrossRef\]](#).
48. Masui A, Haneda H, Ogata Y. Effects of methane hydrate formation on shear strength of synthetic methane hydrate sediments. In: Proceedings of the 15th International Offshore and Polar Engineering Conference; 2005 Jun 19–24; Seoul, Republic of Korea.
49. Deng W, Liang J, Kuang Z, Xie Y, Yan P. Estimation of Gas Hydrate Saturation Regarding the Hydrate Morphology in Hydrate-Bearing Sands in the Qiongdongnan Basin, South China Sea. *Pure Appl Geophys.* 2023;180:2757–73. [\[CrossRef\]](#).
50. Qin XW, Lu JA, Lu HL, Qiu HJ, Liang JQ, Kang DJ, et al. Coexistence of natural gas hydrate, free gas and water in the gas hydrate system in the Shenhua Area, South China Sea. *China Geol.* 2020;2:210–20. [\[CrossRef\]](#).
51. Ye HY, Chen JY, Yao YX, Chen DY, Wu XZ, Li DY, et al. Gas recovery from low-permeability muddy silt gas hydrate reservoirs by depressurization coupled with hot water injection: Impact of hydro-lock effect. *Energy.* 2025;316(1):134413. [\[CrossRef\]](#).
52. Li LJ, Li XS, Wang Y, Qin CZ, Li B, Luo YJ, et al. Investigating the interaction effects between reservoir deformation and hydrate dissociation in hydrate-bearing sediment by depressurization method. *Energies.* 2021;14:548. [\[CrossRef\]](#).